



## CHAPTER 3: MOVEMENT OF SUBSTANCES ACROSS THE PLASMA MEMBRANE



### Characteristics of Molecules that Can or Cannot Diffuse Across the Phospholipid Bilayer

1. **Polar** molecules are **water soluble** and **hydrophilic**.
2. **Non-polar** molecules are **lipid-soluble** (fat-soluble) and **hydrophobic**.
3. Polar molecules dissolve in other polar molecules, and non-polar molecules dissolve in other non-polar molecules.
4. The phospholipid bilayer forms a **barrier** to the diffusion of **hydrophilic** (polar) molecules.
5. In general, **ions** of any size, **most polar molecules** and **most large molecules cannot diffuse** through the phospholipid bilayer; they have to go through the pore proteins.

*Characteristics of some molecules that can and cannot diffuse through the phospholipid bilayer*

Types of molecules that can diffuse through the phospholipid bilayer	Characteristics of the molecules
Water, glycerol, ethanol, urea	<ul style="list-style-type: none"> <li>• Small, uncharged, polar (hydrophilic) molecules</li> <li>• Although the hydrophobic tails of the phospholipids prevent polar water molecules from passing through them, water molecules are <b>small enough</b> to squeeze between the phospholipid bilayer gaps or move through the pores.</li> </ul>
Oxygen	<ul style="list-style-type: none"> <li>• Small, uncharged, non-polar (hydrophobic) molecules</li> </ul>
Carbon dioxide	
Nitrogen	
Benzene	
Fatty acid	
Alcohol	<ul style="list-style-type: none"> <li>• Non-polar, lipid-soluble molecules</li> </ul>
Steroid	
Vitamins A, D, E, K	
Types of molecules that cannot diffuse through the phospholipid bilayer	Characteristics of the molecules
Glucose, sucrose	<ul style="list-style-type: none"> <li>• Large uncharged polar molecules</li> </ul>
Amino acid	
Nucleic acid	
Polysaccharides	<ul style="list-style-type: none"> <li>• Large molecules</li> </ul>
Ions, e.g. $H^+$ , $Na^+$ , $HCO_3^-$ , $K^+$ , $Ca^{2+}$ , $Cl^-$ , $Mg^{2+}$	<ul style="list-style-type: none"> <li>• Charged, regardless of molecule size, as they are highly hydrated, i.e. combined with water molecules</li> </ul>