





# **CHAPTER 4: MATTER AND SUBSTANCE**

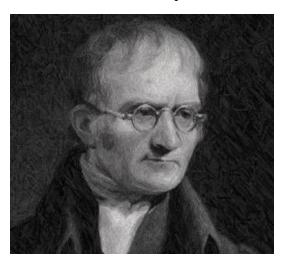


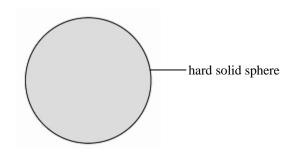
## **Graphic on the Development of the Atom**

Although an atom is so tiny, scientists now have a very good picture of the structure of an atom. But this knowledge has been slowly and patiently developed over the years.

#### John Dalton

The atom is a hard solid sphere. The atom cannot be created or destroyed.

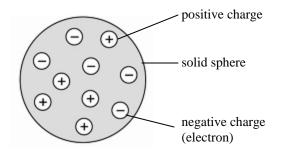




#### J.J. Thomson

The atom is a solid sphere that consists of positively charged solid matter. Negative charges (electrons) are embedded in it. On the whole an atom is neutral.





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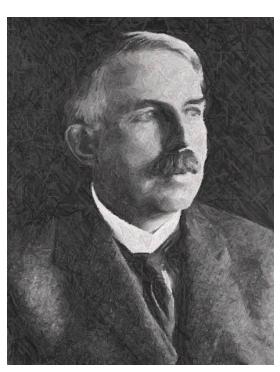


#### **Bohr Rutherford**

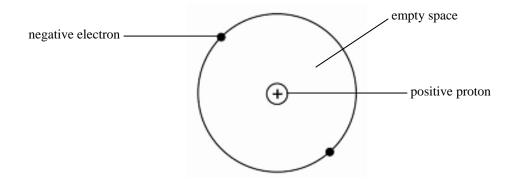
The atom is like a 'solar system'. It has a positively charged nucleus containing protons with negatively charged electrons orbiting a nuclear 'sun'. Most of the atom is empty space.







Ernest Rutherford



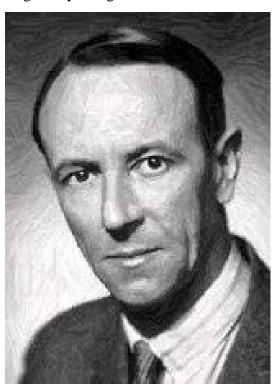
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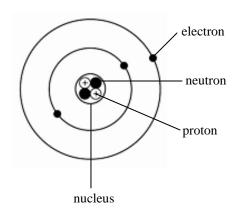




#### **James Chadwick**

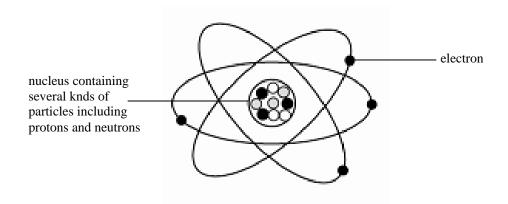
The atom has a nucleus built up from positively charged protons and neutral neutrons. Negatively charged electrons orbit the nucleus in their own shells.





### Present day

The atom has a nucleus built up from protons, neutrons and other subatomic particles. Negatively charged electrons orbit the nucleus in their own shells.



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